

The History of Water Chlorination

Prior to 1908 no municipal water system chemically disinfected water. Waterborne diseases such as typhoid fever and cholera claimed the lives of many Americans. The use of chemicals as a disinfection method was considered illogical and unsafe. John Leal, a physician from New York went against the common beliefs of his time and revolutionized the treatment of water.

John L. Leal graduated from Princeton College in 1880. He went on to graduate from Columbia College of Physicians and Surgeons in 1883. He opened a medical practice in Paterson, New Jersey. In 1899 he became the sanitary adviser to private water companies such as East Jersey Water Company and Jersey City Water Supply Company.

Jersey City had suffered with a contaminated water supply for decades. Tens of thousands of people died from typhoid fever and other waterborne diseases. In 1899 the city contracted with Jersey City Water Supply Company to build a dam on the Rockaway River and create a new water supply. The dam created Boonton Reservoir which had a storage capacity of seven billion gallons. Leal's job was to remove



sources of contamination in the Rockaway River above the reservoir. Water from the reservoir was served to the city on May 23, 1904.

Jersey City claimed that the water was not "pure and wholesome" and filed suit against the Jersey City Water Supply Company. Leal and the company attorney argued to install "other devices" that would do a better job. The judge gave them a little over three months to prove their idea. In May of 1908 Leal decided it was time to add a chemical disinfectant to drinking water. Leal was familiar with some success

CAWWA

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Este informe contiene la información importante! Si usted no entiende este informe, pida que alguien lo traduzca usted.

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Chemicals added to water as a disinfectant had never been attempted in the United States. The public feared chemicals in their food and water. Leal was convinced that adding chlorine to the water was the best way to kill disease causing bacteria. Leal hired George Warren Fuller to design a chlorine feed system to treat 40 million gallons per day. Ninety-nine days later the system was built and operational. The system worked from day one and continued to work without any problems.

Thanks to the courage of people like John Leal and George Fuller many lives were saved. The Chlorination and Fluoridation of water are considered two of the greatest public health achievements of the 20th Century in the United States. The Anniston Water Works is proud to serve our customers with safe, high quality drinking water.

Ed Turner, General Manager

DUR MISSION IS:

- SERVICE by providing high quality drinking water to our customers on demand while maintaining our plants and equipment to facilitate economic growth and development.
- **PROTECTION OF THE ENVIRONMENT AND PUBLIC HEALTH** through responsible wastewater treatment and source water protection
- CONTINUOUS IMPROVEMENT of our processes and personnel to achieve the highest standards of customer satisfaction and to meet or exceed all water and wastewater quality standards.

Important Information to Know about Water

- Substances that may be present in source water include: Microbial contaminates, such as viruses and bacteria, which may come from sewage treatment plants, 0 septic systems, agricultural livestock operations and wildlife.
- Inorganic contaminates, such as salts and metals, which can be naturally occurring, or as result from urban run-off, industrial or domestic wastewater dis-0 charges, oil or gas production, mining or farming.
- Pesticides and herbicides, which may come from a variety of sources such as agriculture, urban storm water run-off, and residential uses, organic chemical 0 contaminates, including synthetic and volatile organic chemicals, which are byproducts of industrial processes and petroleum production, and can also come from gas stations, urban storm run-off, and septic tanks.
- Radioactive contaminates, which can be naturally occurring or be the result of oil and gas production and mining activities. 0
- In order to ensure that tap water is safe. EPA prescribes regulations which limit the amount of certain contaminates in water provided by public water systems. 0 Food and Drug Administration regulations establish limits for contaminates in bottled water, which must provide the same protection for public health.
- Some people may be more vulnerable to contaminants in drinking water than the general population. People who are immuno-compromised such as cancer pa-tients undergoing chemotherapy, organ transplant recipients, HIV/AIDS positive or other immune system disorders, some elderly, and infants can be particularly 0 at risk from infections. Those at risk should seek advice about drinking water from their health care providers. EPA/CDC quidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbiological contaminants are available from the Safe Drinking Water Hotline (800-426-4791). This information is being provided in addition to other information or notices that may be required by law.

,1 - Dichloropropene	2,2 - Dichloropropane	Hexachlorobutadiene	P-Isopropyltoluene	1,1-Dichloroethylene	Ethylbenzene	Beryllium	Selenium
,1,2,2-Tetrachloroethane	Bromobenzene	lsopropylbenzene	Sec - Butylbenzene	1,2,4-Trichlorobenzene	p-Dichlorobenzene	Cadimium	Thallium

List of Non-Detect Substances (Anniston Water Works tested for the following substances in 2014 but none were detected.)

2,2 - Dichloropropane	Hexachlorobutadiene	P-Isopropyltoluene	1,1-Dichloroethylene	Ethylbenzene	Beryllium	Selenium
Bromobenzene	lsopropylbenzene	Sec - Butylbenzene	1,2,4-Trichlorobenzene	p-Dichlorobenzene	Cadimium	Thallium
Bromochloromethane	M-Dichlorobenzene	Tert - Butylbenzene	1,2-Dichloroethane	Styrene	Chromium	Color
Bromoform	MTBE	Trichlorfluoromethane	1,2-Dichloropropane	Tetrachloroethylene	Cyanide	Foaming Agents
Bromomethane	N - Butylbenzene	1,1,1,2-Tetrachloroethane	Benzene	Toluene	Lead	Manganese
Chloroethane	Naphthalene	Trans 1,3 Dichloropropene	Carbon Tetrachloride	trans-1,2-Dichloroethylene	Mercury	Silver
Chloromethane	N-Propylbenzene	0-Dichlorobenzene	Chlorobenzene	Vinyl Chloride	Nickel	Zinc
Dibromomethane	O-Chlorotoluene	1,1,1-Trichloroethane	cis-1,2-Dichloroethylene	Xylenes	Nitrate	Dibromoacetic Acid
Dichlorodifluoromethane	P-Chlorotoluene	1,1,2-Trichloroethane	Dichloromethane	Antimony	Nitrite	Monobromoacetic Acid
	Bramabenzene Bramachlaramethane Bramaform Bramamethane Chloroethane Chloromethane Dibramamethane	Bromobenzene Isopropylbenzene Bromochloromethane M-Dichlorobenzene Bromoform MTBE Bromomethane N - Butylbenzene Chloroethane Naphthalene Chloromethane N-Propylbenzene Dibromomethane D-Chlorotoluene	Bromobenzene Isopropylbenzene Sec - Butylbenzene Bromochloromethane M-Dichlorobenzene Tert - Butylbenzene Bromoform MTBE Trichlorfluoromethane Bromomethane N - Butylbenzene 11.1.2-Tetrachloroethane Chloroethane N - Butylbenzene 11.1.2-Tetrachloroethane Chloromethane N - Butylbenzene 11.1.2-Tetrachloroethane Chloromethane N - Popylbenzene D-Dichlorobenzene Dibromomethane D-Chlorotoluene 1.1.1-Trichloroethane	Bromobenzene Isopropylbenzene Sec - Butylbenzene 12.4-Trichlorobenzene Bromochloromethane M-Dichlorobenzene Tert - Butylbenzene 1.2-Dichlorobenzene Bromoform MTBE Trichlorfluoromethane 1.2-Dichloropropane Bromomethane N - Butylbenzene 1.1.2-Tetrachloromethane 1.2-Dichloropropane Bromomethane N - Butylbenzene 1.1.2-Tetrachloromethane Benzene Chloromethane N - Butylbenzene Trins 1.3 Dichloropropene Carbon Tetrachloride Chloromethane N-Propylbenzene 0-Dichlorobenzene Chlorobenzene Chlorobenzene Dibromomethane D-Chlorotoluene 1.1.1-Trichloroethane cis-1.2-Dichloroethylene	Bromobenzene Isopropylbenzene Sec - Butylbenzene 1.2.4-Trichlorobenzene p-Dichlorobenzene Bromochloromethane M-Dichlorobenzene Tert - Butylbenzene 1.2-Dichlorobenzene Styrene Bromoform MTBE Trichlorfluoromethane 1.2-Dichloroppane Tertachloroethylene Bromomethane N - Butylbenzene 1.11.2-Tetrachloroethane 1.2-Dichloroppane Tetrachloroethylene Bromomethane N - Butylbenzene 1.11.2-Tetrachloroethane Benzene Toluene Chloroethane N - Butylbenzene Trans 1.3 Dichloroppropene Carbon Tetrachloride trans-1.2-Dichloroethylene Chloromethane N-Propylbenzene D-Dichlorobenzene Chlorobenzene Vinyl Chloride Dibromomethane N-Propylbenzene D-Dichlorobenzene Chlorobenzene Vinyl Chloride	Bromobenzene Isopropylbenzene Sec - Butylbenzene 1.2.4-Trichlorobenzene p-Dichlorobenzene Cadimium Bromochloromethane M-Dichlorobenzene Tert - Butylbenzene 1.2.Dichlorobenzene Styrene Chromium Bromochloromethane M-Dichlorobenzene Tert - Butylbenzene 1.2.Dichloropethane Styrene Chromium Bromoform MTBE Trichloromethane 1.2.Dichloropropane Tetrachloroethylene Cyanide Bromomethane N - Butylbenzene 1.1.2-Tetrachloroptene Benzene Toluene Lead Chloromethane N - Butylbenzene 1.1.2-Tetrachloroptenpene Carbon Tetrachloride trans-1.2-Dichloroethylene Lead Chloromethane Naphthalene Trans 1.3 Dichloroptoppene Carbon Tetrachloride trans-1.2-Dichloroethylene Mickel Chloromethane N-Propylbenzene D-Dichlorobenzene Chlorobenzene Vinyl Chloride Nickel Dibromomethane D-Chlorotoluene L1.1-Trichloroethane cis-1.2-Dichloroethylene Xylenes Nitrate

Federal Unregulated Monitoring

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1,2,3-Trichloropropane	1,1-Dichloroethane	Bromochloromethane	Molybdenum	Chromium	Perfluorooctanesulfonic Acid	Perfluorohexanesulfonic Acid	
1,3-Butadiene	Bromomethane	1,4-Dioxane	Cobalt	Chromium-6	Perfluorooctanoic Acid	Perfluoroheptanoic Acid	
Chloromethane	Chlorodifluoromethane	Vanadium	Strontium	Chlorate	Perfluorononanoic Acid	Perfluorobutanesulfonic Acid	



			J /	ANUARI L	ECEMBER 2	2014		
Water Source				Coldwater Spring	Hillabee Reservoir			
Primary Inorganic Substances	Units	MCL	MCLG	Highest Level	Last 12 Months	Violation (Yes/No)	Source of Substance	
Arsenic	ppb	50	-	0.55	Less than 0.5	No	Runoff from orchards; natural deposits; runoff from glass and electronics prod tion wastes	
Barium	ppb	2000	2000	23.7	3.3	No	Discharge of drilling wastes; discharge from metals refineries; erosion of natu deposits	
Fluoride	ppm	4	4	0.760	0.781	No	Erosion of natural deposits; water additive which promotes strong teeth; discharge from fertilizer and aluminum factories	
Sulfate	ppm	500		2.22	20.6	No	Erosion of natural deposits	
Secondary Inorganic Substances	Units	MCL	MCLG	Highest Level	Last 12 Months	Violation (Yes/No)	Source of Substance	
Alkalinity, Total	ppm		-	97.7	10.5	No	Erosion of natural deposits	
Aluminum	ppb	200	-	11.60	170.0	No	Water additive for removing organics; Erosion of natural deposits	
Calcium	ppm		-	22.1	11.6	No	Erosion of natural deposits	
Carbon Dioxide	ppm	-		Less than 1.00	Less than 1.00	No	Erosion of natural deposits	
Chloride	ppm	[250]		2.89	5.36	No	An inorganic constituent in water affecting taste	
Conductance	umhos/ cm			195	96.7	No	Erosion of natural deposits	
Copper	ppb	1000	1000	20.0	4.42	No	Corrosion of household plumbing systems; Erosion of natural deposits	
Hardness, Total (As CaCO ₂)	ppm		-	102	33.4	No	Erosion of natural deposits	
	ppb	300	-	55.6	Less than 10.0	No	Erosion of natural deposits	
Magnesium	ppm		-	11.3	1.09	No	Erosion of natural deposits	
Н	ppm		-	7.7	7.1	No	An indicator of acidity or alkalinity levels of water	
Sodium	ppb		-	1.45	1.60	No	Erosion of natural deposits	
otal Dissolved Solids	ppm	[500]		145	76	No	Erosion of natural deposits	
Disinfection By-Products (at the Plants)	Units	MCL	MCLG		Average	Violation (Yes/No)	Source of Substance	
otal Trihalomethanes (TTHM's)		N/A		Less than 0.5	31	No	Source of Substance By-product of drinking water chlorination	
Ialoacetic Acids (HAA5's)	ppb	N/A N/A	0		34.8		By-product of drinking water chlorination By-product of drinking water chlorination	
× /	ppb							
Disinfection By-Products (in Distribution System)	Units	MCL	MCLG			Violation (Yes/No)	Source of Substance	
Total Trihalomethanes (TTHM's)	ppb	80	0		L	No	By-product of drinking water chlorination	
Haloacetic Acids (HAA5's)	ppb	60	0	Less than 6.00	Less than 6.00	No	By-product of drinking water chlorination	
Haloacetic	aiomethanes Acids (HAA5	(TTHM'S) ar 's) are the su	e the sum of t im of the conc	he concentrations of bromoform, brom entrations of dibromoacetic acid, dichl	oroacetic acid, monobromacetic acid,	and trichloroacetic acid N	vL equal to or less than 60 ppb. MCL equal to or less than 60 ppb.	
Regulated Volatile Chemicals	Units	MCL	MCLG	Highest Level	Last 12 Months	Violation (Yes/No)	Source of Substance	
CE (Trichloroethylene)	ppb 5 0 Less than 0.5		Less than 0.5	No	Discharge from metal degreasing sites and other factories			
cis-1.2-Dichloroethylene	ppb	70	70	Less than 0.5	Less than 0.5	No	Discharge from industrial chemical factories	
Non-Regulated Contaminants Table	Units	MCL	MCLG	Highest Level	Last 12 Months	Violation (Yes/No	Source of Substance	
Fotal Organic Carbon	ppb	Not Re	egulated	0.4 1.77		No	Natural sources	
Radionuclides	Units	MCL	MCLG	Water Sources: Coldwater S	pring and Hillabee Reservoir	Violation (Yes/No	Source of Substance	
Gross Alpha	pCi/l	15	0	Sampling not r	equired in 2014	No	Erosion of natural deposits	
			When gross	alpha particle activity exceeds five po		would be analyzed.	· · · · · · · · · · · · · · · · · · ·	
Turbidity	Units	MCL	MCLG	Highest Level Last 12 Months	Highest Level Last 12 Months	Violation (Yes/No	Source of Substance	
Turbidity	NTU	2 con- secutive		0.11	0.27	No	Erosion of natural deposits and soil runoff	
100% of samples were below the turbidity limits. Turbidity has	no health eff	>0.3 fects. Howe	/ ver, turbidity c	an interfere with disinfection and provi	de a medium for microbial growth. Tu	rbidity may indicate the p	resence of disease-causing organisms. These organisms include bacteria, viru	
				ites that can cause symptoms such as		1		
Lead & Copper Monitoring Units MCL MCLG		Distribution System Violations		Violation (Yes/No	Source of Substance			
Lead ppb 15 0		0		No	Corrosion of household plumbing systems; erosion of natural deposits			
Copper	ppb	1300	1300		0	No	Corrosion of household plumbing systems; erosion of natural deposits	
Lead and copper are metals found in natural depo Lead can cause a variety of adverse health effect in normal physical and mental development in bab potential to cause the following effects from a life Copper is an essential nutrient, required by the b periods of exposure can cause gastrointestinal dis more sensitive than others to the effect of copper	sits as ore s when peo ies and yo time expose ody in very sturbance contamina contacted	s containi ople are ex oung childr sure at lev r small am including r include: A	ng other el xposed to in ren, slight o els above t ounts. How hausea and hould cons Anniston Wa	at levels above the action lev eficits in the attention span, h he action level: stroke and kid rever, EPA has found copper to vomiting. Use of water that e: ult their health care provider. ter Works at 256-241-2000 car	el for relatively short periods o earing, and learning abilities o ney disease; cancer. potentially cause the followin kceeds the Action Level over m n provide you with information	terials or in water s if time. These effect f children, and sligh g health effects whe nany years could cau about your facility's	ervice lines used to bring water from the main to the home. ts may include interference with red blood cell chemistry, delays t increases in the blood pressure of some adults. Lead has the n people are exposed to it at levels above the Action Level. Sh se liver or kidney damage. People with Wilsons disease may b water supply; and the Calhoun County Health Department at 25	

MICROBIOLOGICAL SUBSTANCES TABLE FOR PERIOD JANUARY DECEMBER 2014							
Water Source			Coldwater Spring	Hillabee Reservoir			
Total Coliforms MCL MCLG		Highest Level Last 12 Months		Violation (Yes/No)	Source of Substance		
Not more than 5% of the 70 monthly bacteriological samples taken during the month can test positive for total coliform. No sample can test positive for fecal coliform or E. Coli.	Less than 5%	0	1.4	10%	No	Human and animal fecal waste	



Remember *GREASE* the musical?

Remember that GREASE you poured down the drain? It's not a tune to remember but one you won't likely forget! Dispose of household grease in a proper manner -

not in the sanitary sewer!

A Public Information Message from The Water Works and Sewer Board of the City of Annistor For more information contact the Engineering Department at 256-241-5007.

NLC Service Line Warranty Program

The Water Works and Sewer Board of the City of Anniston is notifying residents of a program provided through the National League of Cities (NLC) Service Line Warranty Program administered by Service Line Warranties of America (SLWA). This program offers a warranty to homeowners that covers repairs or replacements on the outside sewer line that runs from the home to the point of utility responsibility. Property owners are accountable for these repairs, not the utility.



For more information on this program or to enroll, please visit SLWA's web site at www.SLWofA.com

Definitions/Abbreviations Used in this Report							
AL	Action Level	The concentration of a contaminant which triggers treatment or other requirements which a water system must follow.					
MCL	Maximum Contaminant Level	The highest level of a contaminant that is allowed in drinking water.					
MCLG	Maximum Contaminant Level Goal	The level of a contaminant in drinking water below which there is no known or expected health risk.					
MRDL	Maximum Residual Disinfectant Level	The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.					
MRDLG	Maximum Residual Disinfectant Level Goal	The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.					
NS	None Set	No MCL has been set.					
NTU	Nephelometric Turbidity Units	A measure of turbidity. Turbidity has no health effects. However, turbidity can interfere with disinfection and provide a medium for microbial growth. Turbidity may indicate the presence of disease- causing organisms. These organisms include bacteria, viruses, and parasites that can cause symptoms such as nausea, cramps, diarrhea, and associated headaches.					
pCi/L	Picocuries Per Liter	A measure of radioactivity.					
РРМ	Parts per Million or milligrams per liter (mg/L)	What is a PPM? Compares to 8 hours and 45 seconds out of a millen- nium (1000 years).					
PPB	Parts per Billion or micrograms per liter (mg/L)	What is a PPB? Compares to 31 seconds out of a millennium (1000 years).					
SU	Standard Unit	A measure of pH or acidity.					
Π	Treatment Technique	A required process intended to reduce the level of a contaminant in drinking water.					

The Alabama Department of Environmental Management (ADEM), with the approval of the United States Environmental Protection Agency (EPA), issued a statewide waiver on monitoring for asbestos and dioxin. Accordingly, Anniston Water Works was not required to monitor for these during the reporting period. Due to the exceptional quality of raw water at Coldwater Spring, the treatment technique at the Paul B. Krebs Water Treatment Plant employs a variance of the filtration rule which was granted by ADEM.

This report is being furnished to you as required by the Safe Drinking Water Act. We are proud to report that your drinking water is safe and meets all requirements of state and federal regulations.

The United States Environmental Protection Agency maintains a Safe Drinking Water Hotline, 800-426-4791, where you can obtain more information about drinking water.

Water Treatment Process



Drinking water supplied to customers of the Anniston System comes from two sources. Our primary water source is the Coldwater Spring located 7 miles west of Anniston on Tom Burkhart Drive. The Alabama Department of Environmental Management classifies Coldwater Spring as groundwater under the influence of surface water. Water from the spring is treated at the Paul B. Krebs Water Treatment Plant. The statement "under the influence," refers to run off into the uncovered spring pool which is over one acre in size.

Our secondary source of water is the Hillabee Creek Reservoir located 7 miles southeast of Anniston on Jennifer Lane. Hillabee Reservoir is classified as a surface water source. Water from the reservoir is treated at the Earl C. Knowlton Water Treatment Plant located just to the north of the reservoir.

Anniston Water Works has completed a Source Water Assessment for Coldwater Spring and for Hillabee Reservoir. Our assessment has found there is 'Low Susceptibility' to our source waters from elements likely to cause contamination. Our assessment will be updated during 2015. Anniston Water Works also owns the Sam H. Hamner Reservoir located 7 miles east of Anniston near the White Plains Community. No water is currently removed from Hamner Reservoir for use in the system.



Anniston Water Works Board of Directors and Management Personnel Ed Turner, General Manager/CED Rodney Owens, Assistant General Manager Del Ferguson, Assistant General Manager Admin Jimmy D'Dell, Chairman Betty Merriweather, Director Jerome Freeman, Vice Chairman Sam Phillips, Director William Robison, Secretary-Treasurer Ann Welch, Director Melvin Womack, Director ard of Directors of the Anniston Water Works consists of four directors appointed by the City of Anniston and t ars appointed by the Calhoun County legislative delegation. The Directors zerve for a period of six years with r

The Board of Directors of the Anniston Water Works consists of four directors appointed by the City of Anniston and three directors appointed by the Calhoun County legislative delegation. The Directors serve for a period of six years with reappointments being made on a staggered basis so all of the members are not replaced during the same year. Board meetings are held on the third Thursday of each month at eleven o clock in the morning at the Main Office located at 331 Mole Street. Suite 200, Anniston, Alabama. Duestions concerning meetings or requests for additional information should be directed to the General Manager and *Or*. Assistant General Manager during normal business hours (Monday-Friday, 7:30 a.m. to 4:30 p.m.) by calling 256-241-2000.